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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
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10/829,160

04/22/2004

Takamitsu Asanuma

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05/02/2006

OLIFF & BERRIDGE, PLC

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ALEXANDRIA, VA 22320

EXAMINER

NGUYEN, TU MINH

ART UNIT

PAPER NUMBER

3748

DATE MAILED: 05/02/2006

Please find below and/or attached an Office communication concerning this application or proceeding.

<b>Office Action Summary</b>	<b>Application No.</b>	<b>Applicant(s)</b>	
	10/829,160	ASANUMA ET AL.	
	<b>Examiner</b>	<b>Art Unit</b>	
	Tu M. Nguyen	3748	

-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --

**Period for Reply**

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

**Status**

- 1) ☒ Responsive to communication(s) filed on 03 March 2006.
- 2a) ☒ This action is **FINAL**.                      2b) ☐ This action is non-final.
- 3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

**Disposition of Claims**

- 4) ☒ Claim(s) 1 and 4-6 is/are pending in the application.
- 4a) Of the above claim(s) \_\_\_\_\_ is/are withdrawn from consideration.
- 5) ☐ Claim(s) \_\_\_\_\_ is/are allowed.
- 6) ☒ Claim(s) 1 and 4-6 is/are rejected.
- 7) ☐ Claim(s) \_\_\_\_\_ is/are objected to.
- 8) ☐ Claim(s) \_\_\_\_\_ are subject to restriction and/or election requirement.

**Application Papers**

- 9) ☐ The specification is objected to by the Examiner.
- 10) ☒ The drawing(s) filed on 22 April 2004 is/are: a) ☒ accepted or b) ☐ objected to by the Examiner.  
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).  
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).
- 11) ☐ The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

**Priority under 35 U.S.C. § 119**

- 12) ☒ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
- a) ☒ All    b) ☐ Some \*    c) ☐ None of:
1. ☐ Certified copies of the priority documents have been received.
2. ☒ Certified copies of the priority documents have been received in Application No. 09/904,875.
3. ☐ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

\* See the attached detailed Office action for a list of the certified copies not received.

**Attachment(s)**

- |   |   |
|---|---|
| 1) <input type="checkbox"/> Notice of References Cited (PTO-892)                        | 4) <input type="checkbox"/> Interview Summary (PTO-413)                     |
| 2) <input type="checkbox"/> Notice of Draftsperson's Patent Drawing Review (PTO-948)    | Paper No(s)/Mail Date. _____  |
| 3) <input type="checkbox"/> Information Disclosure Statement(s) (PTO-1449 or PTO/SB/08) | 5) <input type="checkbox"/> Notice of Informal Patent Application (PTO-152) |
| Paper No(s)/Mail Date _____   | 6) <input type="checkbox"/> Other: _____                                    |

### DETAILED ACTION

1. An Applicant's Amendment filed on March 1, 2006 and an Applicant's Supplemental Amendment filed on March 3, 2006 have been entered. Claims 4-6 have been added with claim 4 amended in the Supplemental Amendment. Overall, claims 1 and 4-6 are pending in this application.

#### *Drawings*

2. The formal drawings filed on April 22, 2004 have been approved for entry.

#### *Double Patenting*

3. Claims 4-6 are provisionally rejected under the judicially created doctrine of obviousness-type double patenting as being unpatentable over claims 2-4, respectively, of copending Application No. 09/904,875. Although the conflicting claims are not identical, they are not patentably distinct from each other because claim 4 in the instant application is broader in scope than claim 2 of the copending Application.

This is a provisional obviousness-type double patenting rejection because the conflicting claims have not in fact been patented.

4. The nonstatutory double patenting rejection is based on a judicially created doctrine grounded in public policy (a policy reflected in the statute) so as to prevent the unjustified or improper timewise extension of the "right to exclude" granted by a patent and to prevent possible

harassment by multiple assignees. See *In re Goodman*, 11 F.3d 1046, 29 USPQ2d 2010 (Fed. Cir. 1993); *In re Longi*, 759 F.2d 887, 225 USPQ 645 (Fed. Cir. 1985); *In re Van Ornum*, 686 F.2d 937, 214 USPQ 761 (CCPA 1982); *In re Vogel*, 422 F.2d 438, 164 USPQ 619 (CCPA 1970); and, *In re Thorington*, 418 F.2d 528, 163 USPQ 644 (CCPA 1969).

A timely filed terminal disclaimer in compliance with 37 CFR 1.321(c) may be used to overcome an actual or provisional rejection based on a nonstatutory double patenting ground provided the conflicting application or patent is shown to be commonly owned with this application. See 37 CFR 1.130(b).

Effective January 1, 1994, a registered attorney or agent of record may sign a terminal disclaimer. A terminal disclaimer signed by the assignee must fully comply with 37 CFR 3.73(b).

### ***Claim Rejections - 35 USC § 103***

5. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office Action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

6. Claim 1 is rejected under 35 U.S.C. 103(a) as being unpatentable over Seto et al. (Japan Publication 6-117221) in view of Hirota et al. (Japan Publication 6-159037).

As shown in Figures 1 and 9 and indicated in the translated Abstract, Seto et al. disclose a device for purifying the exhaust gas of an internal combustion engine, comprising:

- a NO<sub>x</sub> absorbent (20) arranged in the exhaust system, which carries a catalyst (an alkali metal) for absorbing and reducing NO<sub>x</sub> and an oxidation catalyst (platinum) to absorb oxygen in the exhaust gas, the catalyst absorbing NO<sub>x</sub> when the air-fuel ratio in the surrounding atmosphere thereof is lean and releasing the absorbed NO<sub>x</sub> when the air-fuel ratio is stoichiometric or rich;

- a catalytic apparatus (17) for purifying NO<sub>x</sub> arranged in the exhaust system upstream of the NO<sub>x</sub> absorbent, the catalytic apparatus carries a catalyst (an alkali metal) for absorbing NO<sub>x</sub> when the air-fuel ratio in the surrounding atmosphere thereof is lean and releasing the absorbed NO<sub>x</sub> when the air-fuel ratio is stoichiometric or rich; and

- control means (50, 11) for making the air-fuel ratio in the catalytic apparatus (17) rich to release NO<sub>x</sub> therefrom and purify the released NO<sub>x</sub> by reduction, and making the air-fuel ratio in the NO<sub>x</sub> absorbent (20) rich to release NO<sub>x</sub> from the catalyst of the NO<sub>x</sub> absorbent (20) to purify the released NO<sub>x</sub> by reduction and to release oxygen from the oxidation catalyst and thus to cancel oxygen saturation or contamination on the oxidation catalyst of the NO<sub>x</sub> absorbent.

Seto et al., however, fail to disclose that the NO<sub>x</sub> absorbent also has a function of a particulate filter.

As shown in Figures 1 and 2, Hirota et al. teach that it is conventional in the art to use a particulate filter (10) which carries a NO<sub>x</sub> absorber (26) for absorbing and reducing NO<sub>x</sub>. As clearly illustrated in Figure 2, the particulate filter is a wall-flow device comprising a plurality of partition walls having pores, the partition walls carrying a NO<sub>x</sub> absorber (26) on the exhaust gas upstream side surface for absorbing and reducing NO<sub>x</sub>. A controller in Hirota et al. makes the air-fuel ratio in the particulate filter rich to release NO<sub>x</sub> and active-oxygen from the NO<sub>x</sub>

absorber to purify the released NO<sub>x</sub> by reduction, and to oxidize the particulates trapped on the filter by the released active-oxygen. As indicated in the translated Abstract, the heating in the NO<sub>x</sub> releasing and reduction causes elevated temperature in the filter, which induces the trapped soot to be oxidized easily. It would have been obvious to one having ordinary skill in the art at the time of the invention was made, to have replaced the NO<sub>x</sub> absorbent in Seto et al. with the particulate filter taught by Hirota et al., since the use thereof would have reduced harmful soot emissions in the exhaust gas and saved fuel by inducing soot to combust at an earlier time.

Hirota et al., however, fail to teach or suggest that the trapped soot is oxidized without producing a luminous flame.

Since the heating in the NO<sub>x</sub> releasing and reduction in Hirota et al. causes elevated temperature in the filter, which induces the trapped soot to be oxidized easily, the trapped soot is also oxidized at a lower temperature. One with ordinary skill in the art also recognizes that at a lower temperature, the trapped soot in Hirota et al. is oxidized without producing a luminous flame. Moreover, since the particulate filter in Hirota et al. is operated in an exact manner as that in the pending application to oxidize the trapped soot, there is a similar functionality between Hirota et al. and the pending application. This similar functionality leads one with ordinary skill in the art to realize that the trapped soot in Hirota et al. is also oxidized without producing a luminous flame.

7. Claims 4-6 are rejected under 35 U.S.C. 103(a) as being unpatentable over Dou et al. (U.S. Patent Application 2001/0035006) in view of Hirota et al. (Japan Publication 6-159037).

Re claim 4, as depicted in Figure 17, Dou et al. disclose a device for purifying the exhaust gas of an internal combustion engine, comprising:

Art Unit: 3748

- a particulate filter (6) arranged in the exhaust system (see claim 10);
- a NOx adsorber (4) carries a catalyst for absorbing and reducing NOx, the catalyst absorbing NOx when the air-fuel ratio in the surrounding atmosphere thereof is lean and releasing the absorbed NOx to purify NOx by reduction when the air-fuel ratio is stoichiometric or rich;
- a catalytic apparatus (3) for purifying NOx arranged in the exhaust system upstream of the particulate filter, which catalytic apparatus carries a catalyst (noble metals, barium) absorbing NOx when the air-fuel ratio in the surrounding atmosphere thereof is lean and releasing the absorbed NOx when said air-fuel ratio is stoichiometric or rich (see paragraphs 0039, 0041, and 0042);
- control means (2A or 2B) for making the air-fuel ratio in the catalytic apparatus (3) rich to release NOx from the catalyst of the catalytic apparatus to purify the released NOx by reduction and making the air-fuel ratio in the NOx adsorber (4) rich to release NOx from the catalyst of the NOx adsorber to purify the released NOx by reduction; and
- bypassing means (5A) to make possible the exhaust gas bypass the NOx adsorber and the particulate filter located downstream of the catalytic apparatus (3).

Dou et al., however, fail to disclose that the particulate filter and the NOx adsorber can be combined into one single housing.

As shown in Figures 1 and 2, Hirota et al. teach that it is conventional in the art to use a catalyzed particulate filter (10) which carries a NOx adsorber (26) for absorbing and reducing NOx. As clearly illustrated in Figure 2, the catalyzed particulate filter is a wall-flow device comprising a plurality of partition walls having pores, the partition walls carrying a NOx

Art Unit: 3748

absorber (26) on the exhaust gas upstream side surface for absorbing and reducing NO<sub>x</sub>. A controller in Hirota et al. makes the air-fuel ratio in the catalyzed particulate filter rich to release NO<sub>x</sub> and active-oxygen from the NO<sub>x</sub> absorber to purify the released NO<sub>x</sub> by reduction, and to oxidize the particulates trapped on the filter by the released active-oxygen. As indicated in the translated Abstract, the heating in the NO<sub>x</sub> releasing and reduction causes elevated temperature in the filter, which induces the trapped soot to be oxidized easily. It would have been obvious to one having ordinary skill in the art at the time of the invention was made, to have replaced the NO<sub>x</sub> absorber and the particulate filter in Dou et al. with the catalyzed particulate filter taught by Hirota et al., since the application thereof would have reduced spacing in the device and saved fuel by inducing soot to combust at an earlier time.

Hirota et al., however, fail to teach or suggest that the trapped soot is oxidized without producing a luminous flame.

Since the heating in the NO<sub>x</sub> releasing and reduction in Hirota et al. causes elevated temperature in the filter, which induces the trapped soot to be oxidized easily, the trapped soot is also oxidized at a lower temperature. One with ordinary skill in the art also recognizes that at a lower temperature, the trapped soot in Hirota et al. is oxidized without producing a luminous flame. Moreover, since the particulate filter in Hirota et al. is operated in an exact manner as that in the pending application to oxidize the trapped soot, there is a similar functionality between Hirota et al. and the pending application. This similar functionality leads one with ordinary skill in the art to realize that the trapped soot in Hirota et al. is also oxidized without producing a luminous flame.



Re claim 5, in the modified device of Dou et al., the catalytic apparatus (3) carries the catalyst (noble metals, barium) for absorbing and reducing NO<sub>x</sub>, and during the recovery process of the SO<sub>x</sub> pollution of the catalytic apparatus, the bypassing means (5A) makes the exhaust gas bypass the catalyzed particulate filter (see the last 8 lines of paragraph 0062).

Re claim 6, in the modified device of Dou et al., the catalytic apparatus (3) carries the catalyst (noble metals, barium) for absorbing and reducing NO<sub>x</sub>, and immediately after the finishing of the recovery process of the SO<sub>x</sub> pollution of the catalytic apparatus, the bypassing means (5A) does not make the exhaust gas bypass the catalyzed particulate filter and thus the exhaust gas passes through the catalyzed particulate filter.

### *Response to Arguments*

8. Applicant's arguments with respect to the references applied in the previous Office Action have been fully considered but they are not persuasive.

Re claim 1, in response to applicant's argument that the combination of Hirota et al. and Seto et al. is improper because Hirota et al. fail to disclose or teach a catalyst for absorbing and reducing NO<sub>x</sub> on a location upstream side of a partition wall of a particulate filter (page 8 of Applicant's Amendment), the examiner respectfully disagrees.

As shown in Figures 1-2, Hirota et al. disclose or teach a particulate filter (10) which is a wall-flow device comprising a plurality of partition walls (regions having forward slashes) having pores, wherein the partition walls clearly carry a NO<sub>x</sub> absorber (26) on the exhaust gas upstream side surface of the partition walls for absorbing and reducing NO<sub>x</sub>. Thus, Hirota et al. clearly disclose or teach the claim limitation in dispute.

Re claim 1, in response to applicant's argument that the combination of Hirota et al. and Seto et al. is still improper because the particulate filter (10) in Hirota et al. is most likely being used to replace the upstream catalytic apparatus (17) in Seto et al. (pages 9-10 of Applicant's Amendment), the examiner again respectfully disagrees.

A typical NOx absorbent such as the one (20) in Seto et al. has an optimum operating temperature range that is below the exhaust gas temperature exiting an on-going regenerated particulate filter. Thus, if Seto et al. were to replace the upstream catalytic converter (17) with the particulate filter (10) of Hirota et al., the downstream NOx absorbent (20) in Seto et al. would not purify NOx effectively because the exhaust gas temperature entering the NOx absorbent would be too high and thus, an inadvertent release of the harmful NOx into the atmosphere would occur. Consequently, it would be much more beneficial to replace the NOx absorbent (20) in Seto et al. with the particulate filter (10) taught by Hirota et al. to at least prevent such inadvertent release of harmful NOx into the atmosphere.

Re claim 1, in response to applicant's argument that the combination of Hirota et al. and Seto et al. is still improper because none of the references teaches or suggests a control means to make the air-fuel ratio in the particulate filter rich to release NOx and active-oxygen from the NOx absorber to purify the released NOx by reduction, and to oxidize the particulates trapped on the filter by the released active-oxygen so that the trapped soot is oxidized without producing a luminous flame (pages 11-12 of Applicant's Amendment), the examiner again respectfully disagrees.

As shown in Figure 3B, Hirota et al. teach that when an exhaust gas is made rich of stoichiometry to release the trapped NOx from the NOx absorber (26), a reducing reaction

Art Unit: 3748

occurs in which the trapped NO<sub>3</sub> compounds give up an oxygen atom to become NO<sub>2</sub>. As indicated in the translated abstract, the trapped particulate matter in the filter walls are easily ignited by the heat and obviously, by the excess oxygen given up from the release of trapped NO<sub>x</sub> from the NO<sub>x</sub> absorber. And as indicated above, the heating during the NO<sub>x</sub> release in Hirota et al. causes elevated temperature in the filter, which induces the trapped soot to be oxidized easily at a lower temperature. One with ordinary skill in the art should recognize that at a lower temperature, the trapped soot in Hirota et al. is oxidized without producing a luminous flame. Therefore, Hirota et al. teach or suggest the claimed limitation in dispute.

### *Conclusion*

9. Applicant's amendment necessitated the new ground(s) of rejection presented in this Office Action. Accordingly, **THIS ACTION IS MADE FINAL**. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

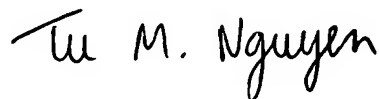
A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the date of this final action.

*Communication*

10. Any inquiry concerning this communication or earlier communications from the examiner should be directed to Examiner Tu Nguyen whose telephone number is (571) 272-4862.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Mr. Thomas E. Denion, can be reached on (571) 272-4859. The fax phone number for the organization where this application or proceeding is assigned is (571) 273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).



TMN

April 29, 2006

Tu M. Nguyen

Primary Examiner

Art Unit 3748